

Cheese Technology Handbook



CHEESE TECHNOLOGY HANDBOOK

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MILK

Composition of milk

The composition of milk varies considerably between different animals. However, only milk from certain animals like a cow, sheep or goat can be used with good results for cheese production.

The main part of cheese in the world is made from cow's milk. The composition of cow's milk is influenced, among others, by feed, breed and stage of lactation. The two main types of protein in milk are casein and whey proteins. Native cow's milk also contains bacteria. Most milk in the world is produced by Holstein breeds.

Table 1. Approximate compositions of milk from different cow breeds.

Cow Breeds	Holstein	Jersey	Guernsey	Ayrshire
Fat	3.3	5.7	5.3	3.8
Protein:	2.8	3.7	3.6	3.1
-Casein	2.2	3.0	2.9	2.5
-Whey	0.6	0.7	0.7	0.6
Protein/ fat ratio	0.85	0.65	0.68	0.82
Lactose	4.6	4.8	4.8	4.9
Ash	0.6	0.8	0.8	0.7

The size of the dry matter components of milk vary in size. The salts have the smallest diameter and the fat, present in globules, and the bacteria have the largest diameters.

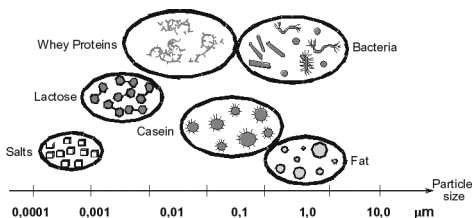


Figure 1. Size of dry matter components in milk.

Determination of milk composition by infrared milk analyzers

Milk composition analyses for fat, protein, lactose and water, can be made by infrared milk analyzers. One of the most used instruments is the Milkoscan (Foss Electric, Hillerod, Denmark). In the instrument, the sample is diluted and homogenized. The mixture then passes through a flow cuvette where the components are measured by their infrared absorption at specific wavelengths.

- Fat at wavelength 5.73 µm
- Protein at wavelength 6.40 µm
- Lactose at wavelength 9.55 µm

The content of water is calculated on the basis of the sum of the values for fat, protein, and lactose plus a constant value for mineral content. The instrument requires exact calibration and must be thermostatically controlled.

Density of milk

The density of milk is correlated to the composition. The usual range is from 8.58 to 8.64 pounds/gallon (1.028 to 1.035 g/mL) for milk. An increased amount of proteins and lactose increase the density, while an increased amount of fat decreases the density value. Thus, cream has lower density than skim milk.

The density changes widely with the temperature, thus all measurements have to be made at the same temperature (usually 60°F/15°C), for results to be compared.

Table 2. Density of milk and cream at 15°C.

Fat (%)	Non-fat solids (%)	Density (Lbs/Gallon)	Density (g/mL)
3.0	8.33	8.61	1.031
3.5	8.60	8.60	1.030
4.0	8.79	8.59	1.029
4.5	8.95	8.58	1.028
5.0	9.10	8.58	1.027
20.0	7.13	8.43	1.010
30.0	6.24	8.36	1.002
40.0	5.35	8.17	0.992

Freezing point

The freezing point of milk is a reliable parameter to check if the milk has been diluted with water (i.e. adulteration). The freezing point of milk from individual cows has been found to vary from 30.94 to 31.03°F (-0.54 to -0.59 °C). Adulteration with water causes the freezing point to increase.

The composition of milk can alter due to physiological or pathological causes (e.g. late lactation and mastitis, respectively), it is termed abnormal milk. The most important change is a fall in lactose content and a rise in chloride content, but the freezing point remains constant.

Fat in milk

The fat in milk is present in fat globules with a diameter of 1-20 μm (0.001-0.02 mm). Because the fat globules have a lower density than the other constituents of milk, they can be separated by centrifugation. Homogenization gives a hard mechanical treatment to milk, and the fat globules then break into smaller fat globules.

During cheese making the fat globules are incorporated into the cheese. Milk is not homogenized before production of the majority of cheese types. Only a small number of cheese types, for example some blue cheese types, are made from homogenized milk.

Proteins in milk

There are two main types of milk proteins – caseins and whey proteins. The caseins are assembled in particles (i.e. casein micelles) with an average diameter of 100 nm (0.0001 mm) while the whey proteins form structures with a size of 1-2 nm. Thus, the whey proteins are small and can easily be separated from the caseins by microfiltration.

The caseins form the backbone structure in cheese and largely contribute to cheese texture. Most cheese types do not contain any whey proteins. In most cheese processes, the whey proteins are separated from the caseins during the curd making.

Note that the casein content rather than the total protein content is the critical parameter with respect to cheese yield. Cheese makers are, therefore, advised to regularly monitor the relative amounts of casein, whey proteins and non-protein nitrogen in their milk.

Acidity of milk

The acidity of a solution depends on the concentration of hydrogen ions $[H^+]$ and hydroxyl ions $[OH^-]$ in it. When the concentration of $[H^+]$ and $[OH^-]$ is equal, the solution is called neutral. The pH is verified from the activity of hydrogen ions $[H^+]$ in a solution. When the pH is:

- Lower than pH 7 - The solution is acidic.
- pH 7 - The solution is neutral.
- Higher than pH 7 - the solution is basic or alkaline.

Native cow's milk is slightly acidic (pH 6.7). Many other foods have lower pH. The pH of yoghurt and cheeses is lower than milk. A difference in pH value of 1 represents a tenfold difference in acidity, i.e. pH 5.5 shows a degree of acidity ten times higher than pH 6.5.

Acidity can also be reported in the titratable acidity (TA). This is based on different measurement methods, in which the total content of free and bound acids is determined. The titratable acidity of fresh milk is 17 (Thörner degrees, °Th), 7 (Soxhlet Henkel degrees, °SH), 15.5 (Dornic degrees, °D) and 0.155 (Per cent lactic acid, % l.a). This is equivalent to an approximate pH of 6.7.

In milk, it is the pH value and not the titratable acidity that controls the processes of rennet coagulation, enzyme activity, bacteria growth, reactions of color indicators, taste, etc.

For most process control purposes, pH is a more useful measurement than titratable acidity. Many cheese makers, however, still use titratable acidity to monitor initial acid development during the first hour after adding the starter culture. For this purpose, titratable acidity is a more reliable indicator because relative to pH measurement, it is more sensitive to small changes in milk acidity.

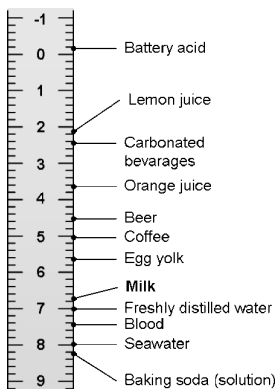


Figure 2. The pH of milk and other common fluids.

Measurement of pH with pH-meter

The pH value is measured by a pH-meter with a combined glass electrode. The system must be carefully calibrated before use.

Determining acidity by titration

Titrateable acidity (TA) of milk is indicated by the number of mL of a sodium hydroxide (NaOH) solution required to neutralize 100 mL of milk, using phenolphthalein as an indicator. Sodium hydroxide solution is added to the milk until the color of the liquid changes from white to a uniform pale red.

The titrateable acidity can be expressed in a variety of units depending on the strength (Molar or N) of the sodium hydroxide solution used for titration.

Table 3. Methods for measuring titrateable acidity, including the amount of milk used for titration with sodium hydroxide (NaOH).

Method/Unit	Milk volume (mL)	Strength NaOH (Molar or N)	Traditionally applied in
Dornic degrees (°D)	100	1/9	Netherlands, France
Percent lactic acid (% l.a.)	Obtained as °D with the result divided by 100		North America, Oceania, UK
Soxhlet Henkel degrees (°SH)	100	0.25	Central Europe
Thörner degrees (°Th)	100 (+200 mL water)	0.1	Northern Europe

When the acidity of cream is determined, the fat content has to be taken into account. If the cream contains 38 % fat and 10 mL of sodium hydroxide solution was used for 100 mL of cream, the acidity ($^{\circ}\text{Th}$) is:

$$10 \times \frac{100}{100-38} = 16.1$$

Preservatives and antibiotics

The growth of lactic acid bacteria may be inhibited by the presence of ordinary antiseptics or antibiotics in the milk. Rapid tests for determination of antibiotics, especially penicillin, in milk have been developed. The Dutch Delvotest P tests for penicillin takes 2.5 hours and penicillin concentrations down to 0.06 I.U./mL can be detected.

TREATMENT OF MILK FOR CHEESE MAKING

Separation and clarification

Centrifuges can be used to separate cream and skim milk. Under the influence of centrifugal forces the fat globules (i.e. cream), which are less dense than the skim milk, move inwards toward the axis of rotation and leave through a central outlet. The skim milk will move outwards and leave through an outer outlet.

During clarification, dense solids like dirt, epithelial cells, leucocytes, corpuscles, bacteria sediment and sludge are separated from the continuous milk phase by centrifugal forces. In modern centrifuges, separation of

cream and clarification is done continuously at the same time in one centrifuge. The dense solids are collected at peripheral discharge slots.

Bactofugation and Bactocatch® processes are also clarification processes. Both processes are used to remove higher number of spores, which otherwise can reduce the cheese quality. This is particularly a problem in Europe. Bactofugation removes 95% and is a centrifugal process, which uses higher forces than separation. Bactocatch® uses microfiltration and achieves about 99% reduction of spore forming bacteria. The spore dense milk phase from Bactofugation and Bactocatch® is UHT treated and added back to the milk.

Standardization

Standardization refers to the practice of adjusting the composition of the cheese milk. Standardization will normally take place automatically before heat treatment to avoid subsequent contamination. The standardization of cheese milk has three separate objectives:

- To maximize economic return from the milk components and the cheese plant investments.
- To maintain consistent cheese quality although the composition of the raw milk changes.
- To meet cheese composition specifications. Specifications can be self imposed (e.g.

low fat cheese) or imposed by government standards for specific cheese types.

Standardization of cheese milk normally requires increasing the protein/fat-ratio. The fat content is decreased by first separating the whole milk into skim milk and cream by means of centrifugal separation. The right amount of skim milk and cream is then mixed to obtain the required fat content of the cheese milk. The whey from the cheese making process also contains a small amount of cream, which can be separated and used in the standardization process of cheese milk.

The protein content can be increased by separation (i.e. membrane filtration) of water from the skim milk, which is later mixed with cream. A second option is to add protein, like skim milk concentrate or low-heat skim milk powder, to the cheese milk.

Because the fat and protein content varies between cheese types, the protein/fat-ratio in cheese milk has to be adjusted. To increase the capacity in the cheese plant, the cheese milk should have a high total solids content, but the protein/fat-ratio should be constant. However, if the total solids content of the cheese milk is too high, the cheese quality will change.

Table 4. Guiding values for protein/fat ratio in cheese milk when producing common cheese types.

Cheese type	Cheese milk protein/fat ratio	Cheese moisture (%)	Cheese fat (%)	Cheese fat in dry matter (%)
Cheddar	0.91	39	31	50.8
Colby	1.03	42	29	50.0
Monterey	1.04	44	28	50.0
Gouda	1.07	43	28	49.1
Edam	1.50	46	22	40.7
Emmental	1.13	40	27	45.0
Havarti	1.19	50	23	46.0
Pizza cheese	1.42	48	20	28.5
Pizza cheese (part skim)	2.20	48	15	28.8
Feta	0.90	55	22	49.8
Cottage cheese	Skim milk	Curd is mixed with "dressing" (cream or non-fat)		

Heat treatment

Heat treatment is applied to cheese milk in order to avoid public health hazards arising from pathogenic microorganisms in the raw milk. The process also increases the shelf-life of the final cheese product. However, cheese can also be made from raw milk that has not been heat treated or undergone thermization.

Table 5. Main types of heat treatments for cheese milk.

Heat treatment	Temperature	Holding time	Purpose
Pasteurization	146°F / 63°C 162°F / 72°C	30 min. 15 sec.	Inactivate and kill pathogenic bacteria
Thermization	145-149°F / 63-65°C	15 sec.	Prevent raw milk spoilage by acid or protease producing bacteria
No	-	-	Results in raw milk cheese which has more flavor

In most countries, regulations require that cheese milk is pasteurized. The pasteurization is intended to only create minimal chemical and organoleptic changes; however pasteurization inactivates enzymes, which contribute to the aroma development during cheese ripening.

To create optimum conditions for cheese making, the cheese milk is pasteurized just before the actual cheese making. If the raw milk has to be stored (cold) more than 24 h before cheese making, there is a risk that certain bacteria will spoil the raw milk. To prevent spoilage during cold storage, thermization is applied. Thermization only kills certain bacteria, and afterwards the milk is still classified as raw milk.

To minimize the risk of failure in the pasteurization process, the system is equipped with an

automatic control system for:

- Pasteurization temperature. The flow is diverted back to the balance tank if the pasteurization temperature is below legal requirement.
- Holding time at pasteurization temperature. The flow of milk is diverted back to the balance tank if the holding time decreases.
- Pressure differential control. The system will activate the flow diversion valve if the pressure on the raw milk side of the regenerator exceeds a set minimum below the pressure on the pasteurized side. This prevents possible leakage of raw milk into the pasteurized milk.

Calculation of holding time

The appropriate tube length for the required holding time can be calculated when the hourly capacity and the inner diameter of the holding tube are known. The velocity profile in the holding tube is not uniform. To ensure that all the milk is sufficiently pasteurized, an efficiency factor must be used. This factor (η) depends on the design of the holding tube, but is often in the range of 0.8-0.9 if the flow is turbulent.

$$(\text{Tube volume, L}) = \frac{(\text{Flow, L/h}) \cdot (\text{Holding time, s})}{3600 \cdot \eta}$$

$$(\text{Tube length, dm}) = \frac{(\text{Volume, L}) \cdot 4}{\pi \cdot (\text{Diameter, dm})^2}$$

To avoid using the second equation, the values for "volumes in stainless steel pipes" can be found in the end of this booklet.

The phosphatase test to test level of heat treatment

In many countries, the phosphatase test is used to determine whether the pasteurization process has been carried out correctly. Phosphatase is an enzyme, and it is inactivated above certain time-temperature combinations. The temperature-time combinations to inactivate important pathogenic bacteria (e.g. Tubercle bacilli) are below the temperature-time combinations for inactivation of phosphatase. Thus, a negative phosphatase test ensures successful inactivation of pathogenic bacteria.

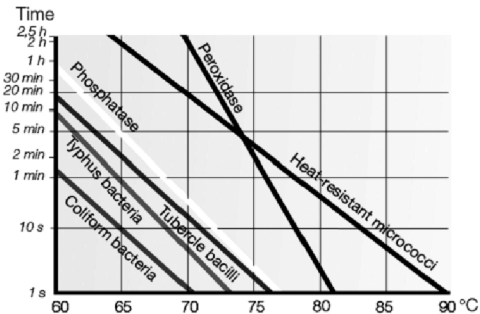


Figure 3. Time-temperature combinations to inactivate certain enzymes and bacteria.

CHEESE TYPES

Cheese varieties can be classified in many different ways based on, among other things, the water content, color, fat content, presence of moulds, region or country of origin. Here we chose to organize cheese types according to process procedures that determine the cheese composition and characteristics.

This results in eight cheese families. The processing of some cheese types is described more in detail in later chapters.

Table 6. Cheese types of different cheese families.

Cheese family	Cheese types	Significant process procedures
Acid-coagulated fresh cheese	Cottage cheese, Quark, Cream cheese	Milk coagulation achieved by acidification (pH 4.6-4.8).
Rennet-coagulated fresh cheese	Queso Blanco, Queso Fresco, Halloumi	Milk coagulation through rennet. Little or no culture is used. The pH is determined by the amount of culture. If no culture is used, the pH remains in the range of 6.5-6.7.
Heat-acid coagulated cheese	Ricotta, Paneer, some varieties of Latin American white cheese	High heat treatment of milk causes denaturation of the whey proteins. Subsequent acidification of the hot milk coagulates both casein and whey proteins. Final pH is normally pH 5.3-5.8.
Soft-ripened cheese	Feta, Camembert, Brie, Blue cheese	Coagulation is primarily by rennet but acidification has considerable influence. Cutting is delayed and done with large knives.
Semi-hard washed cheese	Gouda, Edam, Colby, Brick, Montasio, Oka, Muenster, Danbo, Havarti	Coagulation by rennet. Lactose content is reduced in curd by replacing some whey with water. This limits the acidification to pH 5.0-5.2. The moisture in the cheese is controlled by varying the temperature and time after the wash water was added.
Hard cheese “(Low temp.)”	Cheddar, Monterey Jack, Pasta Filata types.	Milk coagulation by rennet. For Pasta Filata types the curd is worked and stretched in hot water and brine salted. Cheddar types are salted before hooping and pressing.

Cheese family	Cheese types	Significant process procedures
Hard cheese “(High temp.)”	Romano, Parmesan, Swiss	Milk coagulation by rennet. Little acid development before draining. Moisture content is controlled by temperature during renneting and cooking temperature of curd.
Liquid-filled cheese	Cast white cheese, Cast Feta	Renneting of concentrated acidified milk. The concentrate has the same dry matter as the final cheese.

The total solids (TS) content of cheese types varies between 70 % (e.g. Parmesan) and 21 % (e.g. Cottage cheese). The fat content of the cheese is varied by standardization of the cheese milk. This makes it possible to also produce low-fat types of cheeses. The fat content is often given as a percentage of the cheese TS. A fat content of 50 % of TS is written as 50+, 45 % as 45+, etc. The designation “full-cream cheese” is used for cheese 50+.

Table 7. Typical composition (weight %) of some common cheese types.

Cheese family	Cheese type	Moisture	Protein	Total Fat	Fat in DM	Salt	pH
Acid-coagulated fresh cheese	Cottage	80	17	0.4	2	nil	5.0
	Quark	72	18	8	28	1.0	4.5
Rennet-coagulated fresh cheese	Queso blanco	52	23	20	42	2.5	5.8
Heat-acid coagulated cheese	Queso blanco	55	19	20	44	3.0	5.4
	Ricotta	92	11	12	45	<5	5.9
Soft-ripened cheese	Camembert	51	19	24	50	2.1	6.9
	Feta	55	14	21	47	3.5	4.4
	Blue	42	21	29	50	2.5	6.5
Semi-hard washed cheese	Colby	40	25	31	52	0.6	5.3
	Gouda	41	25	27	46	0.8	5.8
	Edam	41	25	27	47	1.0	5.7
	Havarti	43	24	26	47	2.2	5.9
	Munster	42	23	30	51	1.8	6.2
Hard cheese "(Low temp.)"	Cheddar	37	25	33	52	1.8	5.5
	Mozzarella	54	19	22	47	1.0	5.3
Hard cheese "(High temp.)"	Parmesan	29	36	26	36	3.0	5.4
	Swiss	37	28	27	44	1.2	5.6
Liquid-filled cheese	Cast-white	57	16	17	40	4.2	4.6

CHEESE ADDITIVES

Cheese additives are already added to the cheese milk before the rennet coagulation of the milk in the cheese vat. The use of additives can serve various purposes, and not all additives are allowed according to all food legislations.

Calcium chloride

Calcium is naturally present in milk and is crucial to give the coagulum a proper texture. By adding additional calcium, through addition of calcium chloride (approximately 0.02 %), to the cheese milk the coagulation process is improved and the amount of required rennet is reduced. Pasteurization changes the state of calcium in milk. Thus, addition of calcium chloride is especially beneficial if the cheese milk is coagulated directly after pasteurization.

Nitrates

Sodium or potassium nitrate is added at levels of about 0.02 % to Edam, Gouda and Swiss cheese to inhibit growth of gas forming Clostridium bacteria. Addition of nitrates can be avoided if the cheese milk has undergone a Bactocatch® process.

Coloring agents

Cheese colors are added to standardize seasonal changes in color or give additional color to some cheeses such as Cheddar and Cheshire. Traditionally, Annatto cheese color has been used for this purpose. Annatto is a carotenoid similar to vitamin A in structure, but it has no vitamin A activity. Annatto color is red to yellow pigment but it usually appears as orange at $\text{pH} > 6$. At lower pH Annatto gives a red tone, which mostly appears as pink in the cheese. At $\text{pH} < 4.8$ the pink fades and becomes nearly white. Annatto is bleached by light.

Alternatives to Annatto are Beta-carotene, which is often too yellow, Apo-8-carotenal, which has the advantage of not getting lost in the whey, and Paprika.

De-colorants

Certain de-colorants are allowed by some legislation. Goat's milk and sheep's milk naturally do not contain carotenes and appear flat white in color. Cow's milk may be whitened to mimic goat's or sheep's milk by adding titanium dioxide or chlorophyll. Titanium dioxide is a white pigment. Chlorophyll masks the natural yellow color but excessive addition makes the cheese green.

Ripening enzymes

There are many products available to accelerate cheese ripening or to develop a broader flavor profile.

Lipases, also lipolytic enzymes, are added to cow's milk to produce cheese such as Feta, which is traditionally made from goat's or sheep's milk. Goat's milk or sheep's milk, especially goat's milk, contain more natural lipase than cow's milk.

Different cocktails of enzymes from various sources can be added to the milk to accelerate ripening of aged cheese such as Cheddar or Gouda.

CHEESE STARTER CULTURES

The starter culture is added to the cheese milk some time before the rennet enzyme is added to induce the coagulation of the milk.

Types of starter cultures

The type of starter culture used in cheese making influences the end product. The starter culture can contain one type of bacteria (i.e. a single-strain culture) or a mixture of different types of bacteria (i.e. a mixed-strain culture). The types of bacteria can be divided in two main groups according to their preferred temperature of developing:

- Mesophilic bacteria with a temperature optimum between 77 and 104°F (25 and 40°C).
- Thermophilic bacteria, which develop at up to 122°F/50°C.

Mixed-strain cultures often consist of either a cocktail of mesophilic bacteria or thermophilic bacteria, or sometimes a combination of both. Gouda, Manchego, Tilsiter, Cheddar and American varieties are generally based on mesophilic cultures, and Emmental and Gruyère generally on thermophilic cultures. Most starter cultures are mixed-strain but single-strain cultures are sometimes used in production of Cheddar and related types of cheese. Two common types of mixed-strain cultures are described below.

In the broadest terms starter cultures have two purposes in cheese making: (i) to develop acidity through production of lactic acid; and (ii) to promote ripening of the cheese. Lactic acid bacteria (LAB) cultures contribute to both of these functions, while numerous special or secondary cultures are added to help with the second function.

Many cultures do not only produce lactic acid, but also have the ability to form carbon dioxide and aroma components. Carbon dioxide is essential for creating the holes in round-eyed cheeses and supports the openness of granular types of cheese. The mesophilic cultures for Cheddar cheese do not produce carbon dioxide.

Table 8. Two common cheese cultures.

Mesophilic culture	
Bacteria	Comments
Lactococcus lactis ssp cremoris Lactococcus lactis ssp lactis	As a mixed blend these two form the most common mesophilic and homofermtative (no gas production) culture. Used for many low temperature varieties (e.g. fresh cheese, Cheddar, American varieties, etc.)
Thermophilic culture	
Bacteria	Comments
Streptococcus salivarius ssp thermophilus Lactobacillus helveticus	Commonly used for high temperature varieties (Swiss and Italian cheeses). L. helveticus, used to reduce browning in Mozzarella, and to promote proteolysis in Cheddar

Starter systems

The bacteria cultures can be produced in different ways and be in the form of a bulk starter or a direct vat starter. The different types of bulk starter systems can be (i) conventional, (ii) external pH control, and (iii) internal pH control.

Bulk starter systems

Conventional starter is usually made with skim milk at 10% total solids. The medium is heated to 195°F/90°C for 45 minutes to kill any pathogens and bacteriophage. The starter medium is cooled to approximately 78-80°F/25-27°C for mesophilic cultures and 106-108°F/41-42°C for thermophilic

cultures. The bacteria culture to set the starter vat has been kept frozen in an ultra-cold freezer (-40°F/-40°C) before it is thawed. The starter culture is added to the starter vat and agitated. The starter is allowed to grow to a desired pH. Once the desired pH is reached, the tank is agitated or “broken” and cooled to 40°F/4.5°C. The fermentation requires 12-14 hours and the starter culture is added to the cheese vat.

The method of external pH control was developed to produce cultures with maximum acid production rates while maintaining the optimum pH level for acid production rate (activity) in the cheese vat. The media for pH control systems has a total solids level of 5-9 % solids. The media is pasteurized before being cooled to the proper inoculation temperature. After the starter is cooled to set temperature, the culture is added and grown for approximately 8 hours. When the pH drops to 5.8, neutralizer is added until the pH reaches 6.0. Usually aqua-ammonia is used as a neutralizer. This process continues for about 5-7 hours or until the pH no longer drops for at least 45 minutes.

In internal pH control systems, magnesium and phosphate buffers are used in the starter medium to control the pH instead of adding a neutralizing solution. As the pH drops, buffer salts neutralize the acid to maintain the optimum pH for bacterial growth. The internally buffered system can be used for all types of cheeses. The process time is approximately 12-18 hours until a pH of 5.0-5.4.

Direct vat starter (DVS)

Another method of achieving the desired pH in cheese varieties is to use a direct vat starter (DVS). For DVS, bacteria has been concentrated, frozen, packaged, and shipped to the cheese maker. The starter culture is then taken from the freezer and added directly to the cheese vat.

Selecting starter types

Different starter cultures and starter systems are used in cheese production. For example, often a bulk starter system is used in combination with a DVS, which is added directly to the cheese vat. DVS is, however, often more expensive than bulk starters. In emergencies, bulk starter can also be replaced by a DVS. Also when phage problems occur, a simple method of improving the situation is to add some DVS.

Starter problems

Some general problems that occur with starter cultures are inconsistent parameters that affect how the culture grows. For example, if the cooling comes on too soon, while the culture is growing in the starter vat, the starter will not have enough cells to produce the amount of acid to achieve the proper specifications. If the culture is stored in a freezer that fluctuates in temperature, the water crystals change in size and crush the bacteria cells. In general, close observation of critical control points is essential in producing good starter and excellent products.

CHEESE PROCESSING

General processing steps

The feature common to all cheese making is that the fluid milk coagulates and forms a solid gel (i.e. coagulum) after acidification and/or addition of rennet enzyme. It is the casein proteins of the milk that react due to acidification and rennet addition, and create a ridged network within the coagulum. Fat globules and water get entrapped in the ridged network of the casein proteins.

When the coagulum is cut into slabs or cubes, the pieces of coagulum will contract, which, in turn, causes whey to expel, so-called syneresis. The whey contains whey proteins, lactose, minerals, water and a minor amount of fat. Thus, due to syneresis the originally homogenous milk can be separated into curd grains (cheese) and whey.

Table 9. Transfer of milk and milk components to cheese and whey during curd making.

→ Cheese milk →	10-15 % of total milk	→ Cheese
	89-94 % of fat	
	74-77 % of proteins	
→ Cheese milk →	85-90 % of total milk	→ Whey
	6-11 % of fat	
	23-26 % of proteins	

Although curd making is an important part of cheese making other processing steps and their parameters also heavily affect the quality of both cheese and whey. The sequence and

purpose of common processing steps for hard and semi-hard cheese are summarized below.

Table 10. General processing steps, and their purpose, during cheese making.

Processing step	Description and purpose(s)
Thermization, pasteurization, bactofugation	Heat-treatment to condition milk or to inactivate pathogenic bacteria. Bactofugation removes bacteria.
Separation, standardization	Separation of cream from milk, and then standardization of fat content.
Ripening of milk	Addition of starter culture to ensure culture activity and acid production before rennet is added. Some acidity aids renneting.
Setting milk	Rennet is added to induce coagulation of the cheese milk.
Cutting the curd	Cutting of coagulum into curd grains induces whey to be expelled. Proper cutting is extremely important to both quality and yield. Loss of small curd fines to whey should be avoided.
Cooking curd	Is done in production of some cheeses. The combination of heat and the developing acidity (decreasing pH) causes contraction of curd grains and expulsion of whey (including lactose, acid, minerals, salts, and whey proteins). This influences texture and moisture content in final cheese.
Curd washing	Is done in production of some cheeses. Results in leaching of lactose from curd grains. This gives a high moisture cheese (e.g. Danbo, Muenster) but still a low final pH (5.0-5.2). The temperature of water determines the final moisture content of the curd.
Draining	Permanently separates the whey from the curd grains.

Processing step	Description and purpose(s)
Curd handling (including molding, pre-pressing and cheddaring)	The curd is fused to form a smooth, plastic mass. Curd can be directly dipped into the forms or pressed under the whey (brine or surface salted varieties), pre-pressed in vat or column under whey before draining (e.g. Gouda and Swiss), or curd is kept warm in the vat or drain table and allowed to ferment to pH 5.2 -5.4 (e.g. Cheddar, Pasta Filata).
Pressing	Shape the cheese and close up the body. Little or no pressures (e.g. soft cheese) up to high pressures (e.g. firm Cheddar cheese) are applied. Mechanical openings may be reduced by vacuum treatment.
Salting	Salting can be done through: [i] addition to curd before pressing, [ii] surface salting of curd block after pressing, [iii] immersion of curd block in salt brine. Influences flavor, moisture content, texture, and decreases starter activity.

Cheese yield equations

Calculation of cheese yields is important to monitor the efficiency of the cheese process between batches. The actual yield (Y_A , %) is calculated by:

$$Y_A = \frac{\text{Cheese Weight}}{\text{Milk Weight}} \cdot 100$$

The actual yield can conceal inefficiency in recovery of fat and protein. To better monitor the recovery of fat and protein, a moisture and salt adjusted cheese yield (MSACY, %) can be calculated. The actual moisture and salt contents of the cheese are then related to reference/target levels of moisture and salt.

$$\text{MSACY} = \frac{(\text{AY}) [100 - (\text{actual moisture} + \text{actual salt})]}{[100 - (\text{reference moisture} + \text{reference salt})]}$$

The most widely used equation for predicting cheese yield is VanSlyke (VS) formula. The VS formula exists for different cheese types.

$$VS = \frac{[(F \cdot \text{fat in milk}) + (\text{casein in milk} - A)] \cdot B}{[1 - (\text{actual moisture} / 100)]}$$

where F is the coefficient for fat recovery, A is a coefficient for casein loss, and B is a coefficient to account for cheese solids nonfat. For Cheddar F, A and B are 0.93, 0.1 and 1.09. For low-moisture Mozzarella F, A and B are 0.86, 0.36 and 1.09.

Typical cheese equipment

Cheese vat

Designed to provide stirring of milk when mixing with rennet, temperature control during milk setting/coagulation, cutting and stirring of curd, partial whey drainage, and cooking and washing of curd. Can be designed with horizontal/vertical single/twin shafts.

Draining belts and conveyors

Multi level belt designed to provide whey drainage, residence time and cheddaring for the production of matted curd for Cheddar. The draining conveyor also provides stirring capabilities for the production of granular curd for Pasta Filata cheese types.

Cheese block forming towers for Cheddar

The milled and salted curd chips are drawn by vacuum to the top of a tower. The curd fuses into a continuous columnar mass. Regular blocks of identical size are automatically

guillotined, ejected, and bagged. No subsequent pressing is needed.

Cooker-stretcher for Pasta Filata

Used in the manufacture of Pasta Filata type cheeses. Designed to give hard mechanical treatment and cooking of curd grains after milling.

Pre-pressing vats

The whey is drained off in the batch-operated pre-pressing vat, and the curd is pre-pressed before being portioned, moulded and finally pressed. The system is used for many European-type hard and semi-hard cheeses (e.g. Danbo, Edam).

Continuous drainage systems

The system consists of buffer tanks and columns. The curd/whey mixture is continuously pumped from the tank to the top of the column. The drainage column can be round or rectangular-shaped to suit the cheese to be produced. Due to gravity forces, the curd grains form a cohesive mass. In the bottom of the column, cheese blocks are formed. The system is used for many European-type hard and semi-hard cheeses (e.g. Emmental, Gouda, Havarti) before final pressing.

Cheese presses

Designed to perform final pressing of cheese blocks (e.g. Gouda, Emmental, Danbo) in various moulds.

Curd recovery unit

Designed to collect cheese fines from the whey stream after the cheese has been drained.

Processes for Cheddar cheese types

Cheddar cheese

Depending on quality and type (full fat or low fat), Cheddar cheese contains 35-43% moisture and 20-60% fat in dry matter. In some countries, a legal limit for fat in dry matter of Cheddar cheese has been set. A common manufacture procedure involves:

- Standardization of milk to protein/fat-ratio of approximately 0.91
- Pasteurization and then cooling down to inoculation temperature (86-91°F/30-33°C) before adding starter.
- Addition of starter. The ripening time is usually 20-60 minutes.
- Addition of color and/or calcium chloride. Often the color is diluted in water before addition to the milk.
- Rennet enzyme is added to induce coagulation of the milk. The type of rennet will influence the yield and flavor profile of the cheese further down the process. Renneting time is 30-50 minutes.
- Cutting of coagulum using approximately 3/8 inch (95 mm) knives. If the size of cutting is smaller, the moisture content of the final cheese will decrease. The cut curd is gently agitated.

- Scalding for approximately 15 minutes after cutting. The temperature is increased to 99-106°F/37-41°C during 30 minutes (not more than 1°F/5 min) and then held at that temperature until pH is 6.1 (approximately 20-75 min). If the acidity increases too quickly, the temperature may be raised slightly (maximum 104°F/40°C) to retard the culture.
- When the curd pH is 6.0-6.1 (whey pH 6.2-6.3), part of the whey is removed from the vat. The rest of the whey is removed when the curd enters the mechanical cheddaring equipment. The matted curd is finally milled into small pieces, or chips.
- In the end of the cheddaring equipment, salt is added to the milled curd. The final salt content of the cheese should be about 1.5-2.5%. For 1.7% of salt in the final cheese, the required amount of salt added to the curd is 2.5% of the estimated yield.
- Blocks of curd are formed under vacuum in a blockformer tower. In the lower part of the tower each block is packed in plastic foil.
- The packed cheese is stored for curing. Cold curing (41-46°F/5-8°C) produces the best cheese but ripening is slow. Warm cured cheese (50-61°F/10-16°C) develops flavor rapidly but quality control is more difficult.

Process for pizza cheese

Pizza cheese or Pasta Filata cheese types are characterized by a fermentation of the drained curd to pH 4.9-5.2 and then followed by a process where the curd is stretched in hot

water. This plasticizes the curd and gives the final cheese its characteristic fibrous structure and melting properties. Main steps are:

- Milk is typically standardized to protein/fat-ratio 2.2 and then pasteurized.
- The culture is added and the milk is pre-ripened for 15-45 minutes. Modern cultures contain a strain of *Streptococcus thermophilus* often in combination with strains of *Lactobacillus delbrueckii* or *Lactobacillus helveticus*.
- Rennet is added and the milk is let to coagulate at 90-100°F/32-37°C before cutting into curd grains (5-15mm).
- The cut curd is gently stirred for 10-15 minutes before they are drained on a conveyor and pH is 5.3-5.2.
- At pH 5.0-5.1, hot water (167-185°F/75-85°C) is added. The curd is then stretched in a cooker (140-149°F/60-65°C) and formed into cheese blocks.
- The cheese is cooled in chilled water and immersed in brine (10% NaCl) at 60°F/15°C before packing in sealed bags.

Processes for semi-hard cheese

Colby

Colby cheese is characterized by high moisture, open texture, soft body and short curing. The production procedure for Colby is the same as for Cheddar until the correct acidity is attained for dipping. At this time, the final acidity of Colby is adjusted by washing to remove lactose and

acid, while in Cheddar manufacture lactose is removed during Cheddaring. Colby cheese has higher moisture (40-49 %) and a softer body than Cheddar, and never attains the sharp character of Cheddar.

- Standardize milk to protein/fat-ratio 0.96, pasteurize and cool to 88°F/31°C before adding starter.
- Starter is added and milk is ripened for approximately 1 h.
- Color and calcium chloride is added to the milk, then rennet enzyme.
- The formed gel is cut using 3/8" (9.5mm) knives when curd is firm. The curd is gently agitated.
- Cooking is started approximately 15 minutes after cutting. Temperature is slowly increased to 97-102°F/36-39°C. The temperature is held until whey pH is 6.2-6.3.
- When whey pH is 6.2-6.3, the whey is drained down to the level of the curd in the cheese vat.
- Water (59°F/15°C) is added until the curd-water mixture is 79°F/26°C. Stirring is performed when adding water and for an additional 15 minutes. If the wash water is below (59°F/15°C), less water is added. Colder water produces a cheese with higher moisture content.
- The curd is added to mechanical conveying belts and salted. The salt is left to be dissolved in the curd for 15 minutes before hooping.

- The curd is formed into blocks and vacuum packed by a blockformer tower.
- Curing at 45-55°F/7-13°C for 1-3 months.

Gouda

Gouda cheese originated in the Netherlands and is similar to Edam. Normally Gouda has a higher fat content than Edam but fat in dry matter can vary from 20-60%. Gouda is made in round or block forms and the cheese varies in weight from 600g to 20kg. A gas-forming culture is used to induce eye formation.

- Milk is standardized to a protein/fat-ratio of approximately 1.07 and then pasteurized.
- Annatto color can be added to the milk to standardize color.
- Starter culture is added. The main part of used cultures is mixed multi-strains originating from *Lactococcus lactis* and *Leuconostoc mesenteroides*. The milk is pre-ripened for 15-45 minutes. The acidification continues during coagulation, cutting and stirring.
- Coagulation of the milk is induced by rennet addition. The curd is cut into 0.5-1.0 cm cubes. The curd is stirred in whey for up to 30 minutes. Whey pH should be approximately 6.4.
- One-third of the whey is discharged and water (20-25% of the amount of milk) is slowly added at 140°F/60°C to give final temperature of 95-102°F/35-39°C. Stirring is continued for approximately 15 minutes.
- The curd is pre-pressed in a tower or a vat. The pH after pressing should be 5.3-5.5.

- The cheese blocks are immersed in 20% salt brine. The pH should be 5.15 -5.25.
- The cheeses are coated with wax or plastic, and then incubated at 43-68°F/6-20°C for 5-52 weeks. The pH of Gouda cheese increases during ripening. The pH after 8 weeks should be 5.3-5.5.

Process for Cottage cheese

Cottage cheese is a soft, un-ripened white cooked curd from skim milk. The curd is eventually blended with a dressing that traditionally contains cream and salt, but also can have other ingredients. The fat content of the dressing may be reduced to yield a low-fat product. The curd is manufactured by using short (4-5 h), medium (6-8 h) or long (8-16 h) set times. The set time is dependent upon the rate of inoculation and the temperature of the milk. Important steps are:

- The skim milk is gently pasteurized. Denaturation of whey proteins will harm the texture development of the curd.
- The milk is cooled to setting temperature. For short set 90-93°F/32-34°C is used and for long set 77-82°F/25-28°C.
- The culture and perhaps a small amount of rennet are added to the milk.
- The coagulated milk is cut in pieces of 10-14mm when pH is 4.7-4.8.
- The curd is heated for 10-30min and then cooked at 122-140°F/50-60°C for 70-130min. The time is longer at lower temperatures.

- The curd is drained and washed at 36-39°F/2-4°C before the dressing is added.

Processes for Cast cheese types

Cast cheese refers to the process when the cheese milk has the same, or nearly the same, total solids as the final cheese. This can be done by using membrane filtration (UF) to separate water from the cheese milk before acidification and rennet coagulation. The drive behind the development of Cast cheese has been a more continuous process and an increased yield compared to conventional cheese making. The total solids of the cheese milk, decrease the amount of expelled whey and the whey proteins will also become incorporated into the cheese. This results in a higher yield, but also some negative effects, such as change in consistency, changes in some functional properties, and slower ripening.

The development of UF based cheese making processes has been very successful in certain cases. Cast Feta is one of the very successful examples. It has, however, also been proven that for semi-hard and hard cheese types the high content of whey proteins causes problems which makes it difficult, in some cases impossible, to produce a similar quality cheese as traditionally produced cheese.

Brine salting

The salt brine for salting of semi-hard cheeses usually has a concentration 16-25% sodium chloride. The time for the cheese being

immersed in the salt brine depends on the size of the cheese block. Guidelines are: 0.5kg cheese needs 20h, 3-5kg cheese needs 24h, and 20kg cheese, 5 days or sometimes several weeks.

New brine should be treated with about 0.1% of calcium chloride to prevent sodium being absorbed by the casein proteins on the surface of the cheese block. If sodium is absorbed, the surface of the cheese will bind more water and become soft and slimy. Brine pH should be adjusted to the pH, normally 5.2-5.6 of the cheese.

Heat-treatment is used to inactivate microorganisms in brine. Brine can also be cleaned regularly by filtration, preferably MF. Brine must be continuously agitated to prevent density fractionation (i.e. lower concentration brine on top) and dilution of the brine around the cheese.

MEMBRANE FILTRATION IN CHEESE PROCESSES

Membrane filtration processes have become widely used in cheese processes, as well as in many other dairy processes. The membrane filtration processes are characterized by the capabilities of separating molecules of different sizes and characteristics. The pore sizes of the membranes determine which molecules/particles of the fluid that can be separated. Depending on the pore size of the membranes, distinction is made between four types of membrane filtration techniques: reverse osmosis (RO), nanofiltration (NF),

ultrafiltration (UF) and microfiltration (MF). Due to the difference in pore sizes, the design and operation of the four membrane filtration processes are somewhat different.

Table 11. Operational characteristics and limiting factors for membrane filtration processes.

Process	Membrane pore size (μm)	Common trans-membrane pressure (bar)	Limiting factors for flux of components
RO	10^{-4} - 10^{-3}	30-80	Osmotic pressure
NF	10^{-3} - 10^{-2}	20-35	Osmotic pressure
UF	10^{-2} - 10^{-1}	1-10	Gel formation, Concentration polarization, Retentate viscosity
MF	10^{-1} -10	<1	Trans-membrane pressure control, Pore plugging

Common definitions

There are some very frequently used definitions/expressions that are used in membrane filtration technology:

- Feed: The solution to be concentrated or fractionated.
- Retentate: The concentrate. The liquid retained by the membranes.
- Permeate: The filtrate. The liquid passing through the membrane.
- Flux: The rate of permeate extraction measured in liters per square meter of membrane surface area per hour ($\text{L}/\text{m}^2/\text{h}$)

alternatively gallons per square foot per day (GFD).

- **Concentration factor:** The volume reduction achieved by concentration, i.e. the ratio of initial volume of feed to the final volume of retentate.
- **Membrane fouling:** Deposition, accumulation and binding of feed components on the membrane surface and/or within the membrane pores. This causes an irreversible flux decline during processing.
- **Concentration polarization:** Due to flux of water/permeate through the membrane the concentration of non-permeating feed components will increase close to the membrane surface. This will create a layer that is not bound to the membrane (i.e. compare membrane fouling). This will result in a reversible flux decline.
- **Diafiltration:** A design to obtain better purification of the retentate. Water is added to the feed during membrane filtration with the purpose to wash out low-molecular feed components (e.g. lactose, minerals) which will pass through the membranes.

Reverse osmosis (RO)

RO is based on very dense membranes, rejecting virtually all soluble substances except water. Thus, RO retains both high- and low-molecular substances (i.e. protein, fat, lactose and dissolved salts). Operating pressures are relatively high in order to overcome the osmotic pressure.

The majority of dairy RO systems are based on spiral wound elements. The operating pressure is 30-80 bar depending on the osmotic pressure of the solution. For applications in the dairy industry the pressure is usually 30-40 bar.

RO is used to concentrate liquids with low solid levels, and can replace evaporation until a point where the osmotic pressure becomes a limitation. The energy costs are much lower for RO since the water does not need to be evaporated and condensed. However, the costs for replacing old membranes must also be considered.

Nanofiltration (NF)

NF is basically a special version of RO where the membrane structure is more open allowing mainly smaller ions (e.g. sodium, chloride) to permeate through the membrane. Since the low-molecular weight salts pass through the membrane, the osmotic pressure difference across the membrane is reduced, which requires a lower operating pressure. Spiral wound elements are normally used for NF processes.

Ultrafiltration (UF)

In UF the membrane is very open in structure, and typically allows salts, sugars, organic acids and smaller peptides to pass through the membrane. The osmotic pressure of such high molecular weight compounds is quite low, and consequently, the process is performed at lower pressures, usually in the range of 1-10

bar, than RO and NF.

UF is a combined separation and concentration process. The limiting factor in reaching high flux rates and well-defined separation is mainly related to the formation of gels on the membrane surface. Operation with high velocities across the membrane surface minimizes this effect.

UF processes are normally based on spiral wound elements. The flow channels of spiral wound elements are narrow. If the solids content of the fluid/retentate is high (i.e. high viscosity), elements with wider spacers can be used. For very dry solids contents the membrane elements have to be in plate-and-frame modules.

Table 12. Indicative values of total solids (TS) contents for different dairy fluids that are obtainable by means of UF plate-and-frame module.

Fluid	Total solids in retentate (%)
Whey (WPC 35-80)	35
Skim milk (pH 6.7)	39
Skim milk (pH 4.5)	22
Whole milk (pH 6.7)	52

Microfiltration (MF)

MF is based on a membrane with a more open structure than UF, allowing most dissolved substances to pass through, while non-dissolved particles, bacteria and fat globules are rejected and remain in the retentate.

Thus, the pore sizes are typically in the range 0.1-10 μm . There are also MF membranes with smaller pore sizes (0.01 μm), which then can

facilitate separation between whey proteins (0.001-0.003 μm) and casein micelles (0.01-0.3 μm).

The key to control the MF process is to keep the trans-membrane pressure as low (1 bar or less) and as uniform as possible. This and high fluid velocities across the membrane surface will prevent formation of a gel layer on the membrane surface during MF.

Ceramic membrane elements have been the most suitable alternative for MF plants, but spiral wound elements are less expensive and are being used more and more.

Applications of membrane filtration

Membrane filtration processes are widely used to concentrate fluids and to separate different components.

Table 13. Overview of membrane filtration applications.

Process	Application
RO	Used primarily for the concentration of whey, either for the purpose of reducing transport costs or for subsequent evaporation and drying.
NF	Desalination of whey, milk or permeate from UF.
UF	Protein standardization of milk and concentration of whey. In combination with diafiltration, removal of lactose and soluble minerals can be increased.
MF	Depending on the membrane pore sizes MF can be differently applied. Protein standardization and separation of whey proteins from milk are becoming more common. Removal of bacteria and spores from cheese milk and cheese brine has been used for a long time in certain geographical areas.

Protein standardization of cheese milk

The centrifugal separator makes it possible to separate fat from milk. By adding back a calculated quantity of milk fat to skim milk, it is possible to standardize the fat content of milk.

By means of UF, whole milk or skim milk can be standardized with regards to the protein content. The milk retentate is later mixed with permeate to obtain the right protein concentration. This makes it possible to better control the cheese process. Furthermore, an increased protein content in the cheese milk increases the efficiency of the cheese plant.

In some countries, MF (small pores) is applied to standardize the cheese milk with regards to casein content instead of total protein content. It is actually the casein content, not the total protein content, which determines the cheese making properties of the milk. In this way the curd making properties can be even better controlled. Furthermore, a high-quality whey stream will be obtained. The whey stream will be free from fat, casein, bacteria, rennet and spores. Thus, this whey is a high-quality raw material for producing whey protein products.

Bacteria and spore removal in cheese milk

Traditionally, the natural content of anaerobic spores like Clostridia, which survive a normal pasteurization, has been controlled by addition of nitrate and other additives to the cheese milk. The nitrate will, as it is broken down, prevent the anaerobic spore from developing and producing gas, which would seriously destroy or damage the product.

As a result of consumer demands for natural products without addition of preservatives, many markets now reject cheese that has been produced with nitrate.

Centrifugal separation, heat treatment and MF can be used to avoid use of preservatives in the cheese. After separation of the cream, the skim milk is treated in the MF plant. The permeate will not contain any spores and the small amount of concentrate, containing the spores, can then be UHT treated. The cream is also subjected to heat treatment prior to mixing back in the MF permeate.

Concentration of cheese milk

Concentration of cheese milk by UF can be done to different degrees. Depending on the degree of concentration, different configurations of the membranes and different types of cheese equipment (e.g. vats, cutting knives) have to be used.

In pre-concentration, the concentration of the standardized cheese milk becomes maximum two times. This can be used for most cheese types, and is followed by the traditional cheese making procedure with the normal cheese making equipment. If the TS of the cheese milk is doubled, it will increase the utilization of the cheese vats and whey draining equipment with 100 %.

In part-concentration, the cheese milk is concentrated 3-5 times and the subsequent cheese making procedure is modified. The batch cheese making vats are replaced with a coagulation system, which works in

a continuous mode. The investment in new equipment and a quality of the cheese, which is very different to traditional cheeses, has made this process not very successful. Depending on cheese types, this process can increase the yield by as much as 10-15%.

If the standardized milk is fully concentrated to the final total solids of the cheese, all whey proteins will be incorporated in the final cheese. To produce viscous milk concentrates, the membranes have to be in a plate-and-frame configuration. This process is used for Cast Camembert, Cast Feta and Cast White cheeses.

Whey processing

Membrane filtration operations play an important role in processing and increasing the value of whey from cheese production. The ability of membrane filtration to both concentrate and separate has made it widely used in whey processing. Especially UF with diafiltration is important to produce whey protein concentrates (WPC) of different qualities. This is described in more detail in chapters ahead.

Cheese brine

The chemical and microbiological quality of brine used for salting of cheese is essential for production of high quality cheese. The brine may contain unwanted microorganisms including pathogenic bacteria, yeast and moulds. Traditionally cheese brine has been subjected to different types of treatment, such as heat treatment, kieselguhr filtration, UV

radiation, and the addition of preservatives. MF is seen as the ideal process for sanitation of cheese brine because it is fairly simple to operate, does not destroy the balance in the composition of the brine, does not produce any large quantities of waste material, and it is reasonable in terms of maintenance costs.

WHEY AND PERMEATE PRODUCTS

Today the whey solution from cheese production is a valuable raw material for production of whey powder as well as high value added products like whey protein concentrate (WPC) and whey protein isolate (WPI). The products produced from whey are used in a wide range of food products (e.g. processed meat, sausages, health foods, beverages, confectionery). The quality of the raw whey solution is different depending on the type of manufactured cheese and the control of the cheese process. It is mainly distinguished between sweet whey, from manufacture of hard and semi-hard cheeses, and acid whey, from production of cottage cheese and quark. The sweet whey is less acidic and has the highest quality.

Membrane filtration processes play a major role in the manufacture of whey and permeate products. The membrane filtration processes are used for concentration, demineralization, and separation of lactose before evaporation and spray drying of the whey/permeate into powders.

The first step in whey treatment is separation of fat and small particles of casein (i.e. cheese curd) from the whey by centrifugal separation. The next steps are pasteurization and then some sort of concentration, and maybe demineralization.

Table 14. Approximate composition of different whey and permeate powders. Levels of moisture and fat are not given.

Products	Proteins (%)	Lactose (%)	Minerals (%)
Sweet whey powder	12-15	>70	<9
Whey protein concentrate (WPC) powder	35-80	5-45	3-8
Whey protein concentrate (WPI) powder	>90	0.5-1.5	<4
Permeate powder	4 (non-protein nitrogen)	>85	7-9
Lactose powder	<0.5	>99	<0.5

Sweet whey powder

Sweet whey powder is the concentrated and dried product of sweet whey from cheese production. Thus, only the water has been removed. RO can be used to increase the total solids content of the fluid before evaporation and spray drying. With the improved efficiency of evaporators, RO has become less important for concentration before spray drying. However, for concentration of sweet whey up to 15-18 % TS, RO can still be competitive.

Whey protein concentrate (WPC)

By using UF membrane filtration to concentrate sweet whey, the majority of lactose and minerals will also be separated from the whey proteins. Evaporation and spray drying of this concentrated fluid will result in a whey protein concentrates (WPC). This is a successful way of increasing the value of the whey. By applying different degrees of diafiltration the relative percentage of whey proteins in the final product can be varied. WPC powders can contain 30-80 % of whey proteins and are labeled WPC30, WPC50, etc.

Whey protein isolate (WPI)

Using extensive diafiltration with UF membranes makes it possible to produce a whey protein isolate (WPI) solution. This solution can then be evaporated and spray dried to produce a WPI powder with 90% protein of total solids.

Permeate powder

Permeate powders are manufactured from the permeate solution from UF of sweet whey. RO, and then evaporation and spray drying are used to produce the powder.

Lactose powder

Lactose powder is produced from the UF permeate of a sweet whey solution. The permeate solution is concentrated to 60-70% total solids through RO, followed by evaporation. The concentrated solution is fed to crystallization tanks, where it is seeded

with lactose crystals and cooled. The cooling causes super-saturation of lactose and the seeds then initiate crystallization of lactose. Depending on the composition of the permeate, it may be necessary to precipitate calcium phosphate before evaporation by addition of sodium hydroxide (i.e. increases pH). This avoids excessive scaling in the evaporator. After the crystallization process has been completed, the lactose crystals are separated from the remaining liquid (i.e. mother liquor) by means of a centrifugal separation. The crystals are further washed and then transferred to a stationary fluid bed lactose dryer.

CLEANING AND SANITIZING

Good cleaning and sanitation routines provide safety and quality of the cheese and whey products. Safety is assured to the consumer eating the cheese. Higher and more consistent quality gives possibility for higher sales and profits.

Active cultures and development of acidity in cheese does not offer adequate protection against pathogenic organisms. It is true that well made products of some cheese types normally offer significant hurdles to most pathogens, however, several pathogens are well known to survive and may even grow under the conditions of cheese manufacture and ripening. Especially, cheeses with minimal acid development (e.g. Queso Blanco) and cheeses which undergo increased pH during curing (e.g. Brie, Camembert) are susceptible to

growth of pathogens. Thus, good hygiene and sanitation is of high importance in the cheese plant, and some very basic routines can be applied.

Table 15. Some basic routines to increase hygiene and food safety.

Location	Guiding routines
Culture room	Separate from milk and cheese processing plant. Positive air pressure. Clean at all times. Restricted access
Drains	Installed water traps. Designed for volumes of whey and wash water during peak periods.
Surfaces	All surfaces clean and sanitizable. All food contact surfaces in stainless steel (exceptions are curing boards and surface for cheese ripening)
Personnel	Clean clothes. High personal hygiene (especially hands). <i>Staphylococcus aureus</i> and fecal coliforms often originate from humans.
Plant environment	Ideally have positive air pressure. Raw milk reception separated from rest of the plant. Regularly check coliform counts of equipment and employees.

Cleaning systems and procedures

Cleaning system and procedure of choice depends on:

- Nature of the soil (fat, protein, milk salts/ stone).
- Water quality (hardness) and availability.
- Surface to be cleaned (rough or smooth).
- Method of application (manual or CIP).
- Environmental concerns and legislation.

Heat-treatment can be used to kill microorganisms but should not be used before food soils have been removed. The soils can undergo reactions and result in products that make cleaning even harder.

Table 16. Properties of food soils.

Food soil	Fluid for solubility	Ease to remove	Heat-induced reactions
Lactose	Water	Easy	Carmelization
Fats	Alkali	Difficult	Polymerization
Proteins	Alkali	Very difficult	Denaturation
Salts	Water/Acid	Easy to difficult	

Stainless steel surfaces can appear smooth to the human eye. From the perspective of proteins and other molecules, a stainless steel surface is a world of mountains and valleys, which give possibility for good adsorption. To clean a stainless steel surface the cleaners must be given time to do their work.

Residues of proteins on surfaces (e.g. in cheese vat) can be detected with fluorescent light. If the surface reflects a bluish/purple light there is a residual protein film.

Cleaning procedures for dairy equipment is based on a number of steps. Differences between procedures do exist with regard to times, temperatures and concentration of chemicals.

Table 17. Main steps for cleaning.

Step	Fluid / medium	Purpose
Pre-rinse	Water	Removes loose soil
Wash	Chlorinated alkaline detergent solution with chelator, (proteases)	Chlorinated alkaline removes fat and protein, detergent provides wettability, and chelator softens water and removes milk stone.
Rinse	Water	Removes loose soil after wash.
Acid rinse	Nitric, phosphoric solution	Complete removal of milk stone and water hardness
Rinse	Water	Rinse with clean water.
Sanitizing	Thermal and chemical sanitizing	Reduce microbial contamination to a safe level.

The exact concentrations of cleaning solution and acid during washing and acid rinse, respectively, are given by the manufacturer. Concentrations are dependent on the water quality (e.g. hardness and purity).

Sanitizing

Sanitizing refers to the reduction of micro-organisms to levels considered to be safe from a public health point of view. This is different from sterilization (destruction of all living organisms) and disinfection (destruction of all vegetative cells, not spores).

Thermal sanitizing

Thermal sanitizing is most effectively and economically done with hot water. It is relatively inexpensive and penetrates into cracks and crevices, however is also a slow process which requires come-up and cool-down time. The time required is determined by the temperature of the water. Legal limits and recommendations for hot-water sanitizing are:

- Minimum 170°F/77°C for 5 minutes. Grade "A" Pasteurized Milk Ordinance (2007 Revision) of the Food and Drug Administration.
- Minimum 185°F/85°C for 15 minutes. Recommendation of the International Dairy Federation.

Chemical sanitizing

The effectiveness of chemical sanitizing is influenced by the surface characteristics of the equipment. Surfaces which contain bio-films cannot be effectively sanitized. Generally, the longer exposure time a chemical sanitizer is in contact with the equipment surface, the more effective the sanitization effect. Temperature is also positively related to microbial kill by a chemical sanitizer. Too high temperatures (> 130°F/55°C) should be avoided because of the corrosive nature of most chemical sanitizers. Furthermore, the activity of a sanitizer, generally, increases with increased concentration. However, concentrations above recommendations do not sanitize better but can give corrosion of equipment.

There are two main types of chemical sanitizer:

- No-rinse food contact surface sanitizer
- Non-food contact surface sanitizer

No-rinse food contact surface sanitizers approved by The Food and Drug Administration (FDA) are found in the Code of Federal Regulations (21 CFR 178.1010).

The chemical sanitizer can be applied as a spray, foam, or as a circulating fluid. If the time between sanitizing and start-up of the food process exceeds four hours, it is recommended to perform re-sanitizing.

TECHNICAL INFORMATION

Velocity in Stainless Steel Pipes

The velocity in stainless steel pipes should not exceed the values (in m/sec or ft/sec) stated below:

Product	Suction Lines				Pressure Lines			
	25 mm	1"	101.6 mm	4"	25 mm	1"	101.6 mm	4"
Milk	1.5	5	2	6.6	2	6.6	2.5	8.2
Cream	1.5	5	1.5	5	2	6.6	2	6.6
Water	3	9.8	3	9.8	3	9.8	3.5	11.5

Volume in stainless steel pipes

Outside Diameter		Inside Diameter		Required Flow/CIP		Volume	
Inches	mm	Inches	mm	USG /min	Litre /sec	USG /ft	Litre/m
1	25	0.87	22.1	9	0.5678	0.0309	0.3838
1.5	38	1.37	34.8	23	1.4511	0.0766	0.9513
2	51	1.87	47.5	43	2.7129	0.1427	1.7722
2.5	63.5	2.37	60.2	69	4.3532	0.2292	2.8465
3	76	2.87	72.9	101	6.3721	0.336	4.1729
4	101.6	3.87	98.3	183	11.5455	0.611	7.5882

Tables showing conversion factors between SI units and other common unit systems.

Example showing use of pressure/stress table:

1450 p.s.i. converted to bar?

Find factor for bar, line p.s.i. = 1

$6.9 \times 10^{-2} \times 1450 \sim 100$ bar

Length

SI-unit	OTHER UNITS			
m	in (inch)	ft (foot)	yd (yard)	mile
1	39.4	3.28	1.09	0.621×10^{-3}
2.54×10^{-2}	1	8.33×10^{-2}	2.77×10^{-2}	15.8×10^{-6}
0.305	12	1	0.333	0.189×10^{-3}
0.914	36	3	1	0.568×10^{-3}
1.161×10^3	63.4×10^3	5.28×10^3	1.76×10^3	1

Area

SI-unit	Other Units		
m ²	in ² (sq. inch)	ft ² (sq. foot)	yd ² (sq. yard)
1	1.55x10 ³	10.8	1.20
0.645x10 ⁻³	1	6.94x10 ⁻³	0.772x10 ⁻³
9.29x10 ⁻²	144	1	0.111
0.836	1.30x10 ³	9	1

Volume

SI-unit	Other Units				
m ³	in ³ (cu. inch)	ft ³ (cu. foot)	yd ³ (cu. yard)	gallon (UK)	gallon (US)
1	61 x 10 ³	35.3	1.31	220	264
16.4x10 ⁻⁶	1	0.579x10 ⁻³	0.214x10 ⁻⁶	3.60x10 ⁻³	4.33x10 ⁻³
2.83x10 ⁻²	1.73x10 ³	1	3.70x10 ⁻²	6.23	7.48
0.765	46.7x10 ³	27	1	168	202
4.55x10 ⁻³	277	0.161	5.95x10 ⁻³	1	1.20
3.79x10 ⁻³	231	0.134	4.95x10 ⁻³	0.833	1

Velocity

SI-unit	Other Units		
m/s	km/h	ft/s	mile/h
1	3.6	3.28	2.24
0.278	1	0.911	0.621
0.305	1.10	1	0.682
0.447	1.61	1.47	1

Density (mass/volume)

SI-unit	Other Units		
kg/m ³	g/cm ³ g/ml	lb/in ³	lb/ft ³
1	10 ⁻³	36.1x10 ⁻⁶	6.24x10 ⁻²
10 ³	1	3.61x10 ⁻²	62.4
27.7x10 ³	27.7	1	1.73x10 ³
16.0	1.60x10 ⁻²	5.79x10 ⁻³	1

Mass

SI-unit	Other Units	
kg	metric tech. unit of mass	lb (pound)
1	0.102	2.21
9.81	1	21.7
1.454	4.63x10 ⁻²	1

Force, Weight

SI-unit	Other Units	
N	kp	lbf (pound force)
1	0.102	0.225
9.81	1	2.21
4.45	0.454	1

Moment of Force

SI-unit	Other Units	
Nm	kpm	lbf x ft
1	0.102	0.738
9.81	1	7.23
1.36	0.138	1

Pressure, Stress

SI-unit	Other Units				
N/m ² Pa(pascal)	bar	kp/cm ² , at (tech. atmosph.)	mmH ₂ O	mmHg torr	lbf/in ² p.s.i.
1	10 ⁻⁵	10.2x10 ⁻⁶	0.102	7.50x10 ⁻³	0.145x10 ⁻³
10 ⁵	1	1.02	10.2x10 ³	750	14.5
98.1x10 ³	0.981	1	10x10 ³	736	14.2
9.81	98.1x10 ⁻⁶	0.1x10 ⁻³	1	7.36x10 ⁻²	1.42x10 ⁻³
133	1.33x10 ⁻³	1.36x10 ⁻³	13.6	1	1.93x10 ⁻²
6.90x10 ³	6.90x10 ⁻²	7.03x10 ⁻²	703	51.7	1
Standard atmosphere (atm), 1atm =101325 N/m ³					

Energy, work, quantity of heat

SI-unit	Other Units				
J, Nm, Ws	kWh	kpm	kcal	Btu (Brit. thermal unit)	ft x lbf (foot pound- force)
1	0.278×10^{-6}	0.102	0.239×10^{-3}	$.948 \times 10^{-3}$	0.738
3.6×10^6	1	0.367×10^6	860	3.41×10^3	2.66×10^6
9.81	2.72×10^{-6}	1	2.34×10^{-3}	929×10^{-3}	7.23
4.19×10^3	1.16×10^{-3}	427	1	3.97	3.09×10^3
1.06×10^3	0.293×10^{-3}	108	0.252	1	779
1.36	0.377×10^{-6}	0.138	0.324×10^{-3}	1.29×10^{-3}	1

Power, heat flow rate

SI-unit	Other Units				
W, Nm/s, J/s	kpm/s	kcal/h	Btu/h	hp (Brit. horse power)	hk (metric horse power)
1	0.102	0.860	3.41	1.34×10^{-3}	1.36×10^{-3}
9.81	1	8.43	33.5	1.32×10^{-2}	1.33×10^{-2}
1.16	0.119	1	3.97	1.56×10^{-3}	1.58×10^{-3}
0.293	2.99×10^{-2}	0.252	1	0.393×10^{-3}	0.399×10^{-3}
746	76.0	641	2.55×10^3	1	1.01
7.36	75	632	2.51×10^3	0.986	1

Thermometric Scales

Celsius and Fahrenheit Degrees

$$^{\circ}\text{C} = 5/9 (^{\circ}\text{F} - 32^{\circ}) \quad ^{\circ}\text{F} = (^{\circ}\text{C} \times 9/5) + 32^{\circ}$$

$^{\circ}\text{C}$	$^{\circ}\text{F}$	$^{\circ}\text{C}$	$^{\circ}\text{F}$	$^{\circ}\text{F}$	$^{\circ}\text{C}$	$^{\circ}\text{F}$	$^{\circ}\text{C}$
-10	14	16	60.8	0	-17.7	52	11.1
-9	15.8	17	62.6	2	-16.6	54	12.2
-8	17.6	18	64.4	4	-15.5	56	13.3
-7	19.4	19	66.2	6	-14.3	58	14.4
-6	21.2	20	68	8	-13.2	60	15.6
-5	23	21	69.8	10	-12.1	62	16.7
-4	24.8	22	71.6	12	-11	64	17.8
-3	26.6	23	73.4	14	-9.9	66	18.9
-2	28.4	24	75.2	16	-8.8	68	20
-1	30.2	25	77	18	-7.7	70	21.1
0	32	26	78.8	20	-6.6	72	22.2
1	33.8	27	80.6	22	-5.5	74	23.3
2	35.6	28	82.4	24	-4.3	76	24.4
3	37.4	29	84.2	26	-3.2	78	25.6
4	39.2	30	86	28	-2.1	80	26.7
5	41	31	87.8	30	-1	82	27.8
6	42.8	32	89.6	32	0	84	28.9
7	44.6	33	91.4	34	1.1	86	30
8	46.4	34	93.2	36	2.2	88	31.1
9	48.2	35	95	38	3.3	90	32.2
10	50	36	96.8	40	4.4	92	33.3
11	51.8	37	98.6	42	5.6	94	34.4
12	53.6	38	100.4	44	6.7	96	35.6
13	55.4	39	102.2	46	7.8	98	36.7
14	57.2	40	104	48	8.9	100	37.8
15	59			50	10		

Saturated Steam Table in °C
(according to Mollier)

Abs press lb/in ²	Temp °C	Enthalpy Steam h _g	Abs press lb/in ²	Temp °C	Enthalpy Steam h _g
0.1	45.45	617.0	2.5	126.79	648.3
0.2	59.67	623.1	3.0	132.88	650.3
0.3	68.68	626.8	3.5	138.19	651.9
0.4	75.42	629.5	4.0	142.92	653.4
0.5	80.86	631.6	4.5	147.20	654.7
0.6	85.45	633.4	5.0	151.00	655.8
0.7	89.45	634.9	5.5	154.72	656.5
0.8	92.99	636.2	6.0	158.08	657.8
0.9	96.18	637.4	6.5	161.21	658.7
1.0	99.09	638.5	7.0	164.17	659.4
1.1	101.76	639.4	7.5	166.97	660.1
1.2	104.25	640.3	8.0	169.61	660.8
1.3	106.56	641.2	8.5	172.13	661.4
1.4	108.74	642.0	9.0	174.53	662.0
1.5	110.79	642.8	9.5	176.83	662.5
1.6	112.73	643.5	10.0	179.04	663.0
1.7	114.57	644.1	12.5	188.92	665.1
1.8	116.33	644.7	15.0	197.36	666.6
1.9	118.01	645.3	17.5	204.76	667.7
2.0	119.62	645.8	20.0	211.38	668.5

Saturated Steam Table in °F
(according to Mollier)

Abs press lb/in ²	Temp °F	Enthalpy Steam h _g	Abs press lb/in ²	Temp °F	Enthalpy Steam h _g
0.08865	32.018	1075.5	95	324.13	1186.2
0.25	59.323	1067.4	100	327.82	1187.2
0.5	79.586	1096.3	105	331.37	1188
1	101.74	1105.8	110	334.79	1188.9
3	141.47	1122.6	115	338.08	1189.6
6	170.05	1134.2	120	341.27	1190.4
10	193.21	1143.3	125	344.35	1191.1
14.696	212	1150.5	130	347.33	1191.7
15	213.03	1150.9	135	350.23	1192.4
20	227.96	1156.3	140	353.04	1193
25	240.07	1160.6	145	355.77	1193.5
30	250.34	1164.1	150	358.43	1194.1
35	259.29	1167.1	160	363.55	1195.1
40	267.25	1169.8	170	368.42	1196
45	274.44	1172	180	373.08	1196.9
50	281.02	1174.1	190	377.53	1197.6
55	287.08	1175.9	200	381.8	1198.3
60	292.71	1177.6	210	385.91	1199
65	297.98	1179.1	220	389.88	1199.6
70	302.93	1180.6	230	393.7	1200.1
75	307.61	1181.9	240	397.39	1200.6
80	312.04	1183.1	250	400.97	1201.1
85	316.26	1184.2	260	404.44	1201.5
90	320.38	1185.3	270	407.8	1201.9

Salt in the Moisture Phase of Cheese (S/M)

Percent Salt

%H ₂ O	2.4	2.3	2.2	2.1	2.0	1.9	1.8	1.7	1.6	1.5	1.4	1.3	1.2	1.1	1.0
40.0	6.00	5.75	5.50	5.25	5.00	4.75	4.50	4.25	4.00	3.75	3.50	3.25	3.00	2.75	2.50
39.0	6.15	5.90	5.64	5.38	5.13	4.87	4.62	4.36	4.10	3.85	3.59	3.33	3.08	2.82	2.56
38.0	6.32	6.05	5.79	5.53	5.26	5.00	4.74	4.47	4.21	3.95	3.68	3.42	3.16	2.89	2.63
37.0	6.49	6.22	5.95	5.68	5.41	5.14	4.863	4.59	4.32	4.05	3.78	3.51	3.24	2.97	2.70
36.0	6.67	6.39	6.11	5.83	5.56	5.28	5.00	4.72	4.44	4.17	3.89	3.61	3.33	3.06	2.78
35.0	6.86	6.57	6.29	6.00	5.71	5.43	5.14	4.86	4.57	4.29	4.00	3.71	3.43	3.14	2.86
34.0	7.06	6.76	6.47	6.18	5.88	5.59	5.29	5.00	4.71	4.41	4.12	3.82	3.53	3.24	2.94
33.0	7.27	6.97	6.67	6.36	6.06	5.76	5.45	5.15	4.85	4.55	4.24	3.94	3.64	3.33	3.03
32.0	7.50	7.19	6.88	6.56	6.25	5.94	5.63	5.31	5.00	4.69	4.38	4.06	3.75	3.44	3.13
31.0	7.74	7.42	7.10	6.77	6.45	6.13	5.81	5.48	5.16	4.84	4.52	4.19	3.87	3.55	3.23
30.0	8.00	7.67	7.33	7.00	6.67	6.33	6.00	5.67	5.33	5.00	4.67	4.33	4.00	3.67	3.33

Conversion Table

1 inch	x	2.5400	= cm
1 foot	x	0.3048	= m
1 yard	x	0.9144	= m
1 mile	x	1609.0000	= m
1 square inch	x	6.4520	= cm ²
1 square foot	x	0.0929	= cm ²
1 square yard	x	0.8360	= cm ²
1 acre	x	4086.8000	= cm ²
1 cubic inch	x	16.3900	= cm ²
1 cubic foot	x	28.3200	= litre
1 pint (liquid UK)	x	0.5680	= litre
1 pint (liquid US)	x	0.4730	= litre
1 UK quart	x	1.1360	= litre
1 US quart	x	0.9460	= litre
1 US gallon	x	3.7850	= litre
1 UK gallon	x	4.5500	= litre
1 ounce	x	28.3500	= g
1 lb	x	0.4540	= kg
1 short ton	x	907.1800	= kg
1 long ton	x	1016.0600	= kg
1 lb per sq. inch	x	0.0700	= kg/cm ²
<hr/>			
1 cm	x	0.3940	= inch
1 m	x	3.2810	= foot
1 m	x	1.0936	= yard
1 km	x	0.6213	= mile
1 cm ²	x	0.1550	= square inch
1 m ²	x	10.7640	= square foot
1 m ²	x	1.1970	= square yard
1 hektare	x	2.4711	= acre
1 cm ³	x	0.0610	= cubic inch
1 m ³	x	35.3200	= cubic foot
1 litre	x	1.7600	= pint (liquid UK)
1 litre	x	2.1100	= pint (liquid US)
1 litre	x	0.2640	= US gallon
1 litre	x	0.2200	= UK gallon
1 g	x	15.4320	= grains
1 kg	x	2.2046	= lb
1 tonne	x	1.1023	= short ton
1 tonne	x	0.9842	= long ton
1 kg/cm ²	x	14.2200	= lb per sq. inch

$^{\circ}\text{C} = 5/9 (^{\circ}\text{F} - 32^{\circ})$

$^{\circ}\text{F} = 9/5 (^{\circ}\text{C} + 32^{\circ}) + 14.22$

Decimal Equivalent Chart

$1/64 = .0156$	$33/64 = .5156$
$1/32 = .0313$	$17/32 = .5312$
$3/64 = .0469$	$35/64 = .5469$
$1/16 = .0625$	$9/16 = .5625$
$5/64 = .0781$	$37/64 = .5781$
$3/32 = .0937$	$19/32 = .5937$
$7/64 = .1094$	$39/64 = .6094$
$1/8 = .125$	$5/8 = .625$
$9/64 = .1406$	$41/64 = .6406$
$5/32 = .1562$	$21/32 = .6562$
$11/64 = .1719$	$43/64 = .6719$
$3/16 = .1875$	$11/16 = .6875$
$13/64 = .2031$	$45/64 = .7031$
$7/32 = .2187$	$23/32 = .7187$
$15/64 = .2344$	$47/64 = .7344$
$1/4 = .25$	$3/4 = .75$
$17/64 = .2656$	$49/64 = .7656$
$9/32 = .2812$	$25/32 = .7812$
$19/64 = .2969$	$51/64 = .7969$
$5/16 = .3125$	$13/16 = .8125$
$21/64 = .3281$	$53/64 = .8281$
$11/32 = .3437$	$27/32 = .8437$
$23/64 = .3594$	$55/64 = .8594$
$3/8 = .375$	$7/8 = .875$
$25/64 = .3906$	$57/64 = .8906$
$13/32 = .4062$	$29/32 = .9062$
$27/64 = .4219$	$59/64 = .9219$
$7/16 = .4375$	$15/16 = .9375$
$29/64 = .4531$	$61/64 = .9531$
$15/32 = .4687$	$31/32 = .9687$
$31/64 = .4844$	$63/64 = .9843$
$1/2 = .5$	$1 = 1$

USEFUL WEBSITES

Public universities and institutes

American Dairy Science Association,
www.adsa.org

South Dakota State University,
www.sdstate.edu

University of California-Davis,
www.drinc.usdavis.edu

University of Guelph,
www.foodscience.uoguelph.ca

Munich Technical University, www.tum.de

Utah State University, www.usu.edu

WI Center for Dairy Research
University of WI – Madison, www.cdr.wisc.edu

Industry associations

American Cheese Society,
www.cheesesociety.org

Dairy Council of California,
www.dairycouncilofca.org

Dairy Council of WI, www.dcwnet.org

International Dairy Foods Association,
www.idfa.org

North Central Cheese Industry Association,
www.northcentralcheese.org

US Dairy Export Council, www.usdec.org

Wisconsin Cheese Makers Association, www.wischeesemakersassn.org

Ingredient suppliers

Cargill, www.cargill.com

Chr. Hansen, www.Chr-Hansen.com

Danisco, www.danisco.com

DSM, www.dsm.com

Processing equipment

Tetra Pak, www.tetrapak.com

Dairy market news

Cheese Market News,
www.cheesemarketnews.com

Cheese Reporter, www.cheesereporter.com

Dairy Reporter, www.dairyreporter.com

Dairy Foods Magazine, www.dairyfoods.com

Exhibitions

Worldwide Food Expo,
www.worldwidefood.com

Anuga, www.anuga.com

CHEESE MAKING GLOSSARY

Acid curd

The custard-like state that milk is brought to when a high level of acidity is created. The acidity is produced by the activity of starter culture bacteria, and it precipitates the milk protein into a solid curd.

Acidity

The amount of acidity (sourness) in the milk. Acidity is an important element in cheese making and it is produced by cheese starter culture bacteria.

Aging

A step in cheese making in which the cheese is stored at a particular temperature and relative humidity for a specified amount of time in order to develop its distinct flavor.

Albuminous protein

Protein in milk which cannot be precipitated out by the addition of rennet. Albuminous protein, or whey protein, remains in the whey and is precipitated by high temperatures to make ricotta.

Bacteria

Microscopic unicellular organisms found almost everywhere. Lactic acid-producing bacteria are helpful and necessary for the making of quality hard cheeses.

Bacteria linens

A red bacteria which is encouraged to grow on the surfaces of cheeses like Brick or Limburger to produce a sharp flavor.

Bacterial-ripened cheese

A cheese upon whose surface bacterial growth is encouraged to develop in order to produce a distinct flavor. Brick and Limburger are examples of bacterial-ripened cheeses.

Cheese color

A coloring added to the milk prior to renneting which will impart various shades of yellow to the cheese. Most coloring is a derivative of the annatto tree.

Cheese Salt

A coarse flake salt. Salt not iodized is the most desirable type to use in cheese making.

Cheese starter culture

A bacterial culture added to milk as the first step in making many cheeses. The bacteria produced an acid during their life cycle in the milk. There are two categories of starter culture: mesophilic and thermophilic.

Cheese wax

A pliable wax with a low melting point which produces an airtight seal which will not crack. Most hard cheeses are waxed.

Clean break

The condition of the curd when it is ready for cutting. A finger or thermometer inserted into the curd at a 45 degree angle will separate the curd firmly and cleanly if the curd has reached that condition.

Cooking

A step in cheese making during which the cut curd is warmed to expel more whey.

Curd

The solid custard-like state of milk achieved by the addition of rennet. The curd contains most of the milk protein and fat.

Cutting the Curd

A step in cheese making in which the curd is cut into equal-sized pieces.

Draining

A step in cheese making in which the whey is separated from the curd by pouring the pot of curds and whey into a cheesecloth-lined colander.

Drip tray

A tray which is placed under a mold during the pressing of a cheese. The drip tray allows the whey to drain into a sink or container.

Homogenization

A mechanical breaking up of the fat globules in milk so that the cream will no longer rise in the milk.

Lactic acid

Acid created in milk during cheese making. Cheese starter culture bacteria consume the milk sugar (lactose) and produce lactic acid as a byproduct.

Lactose

The sugar naturally present in milk. Lactose can constitute up to 5 percent of the total weight of milk.

Milling

A step in cheese making during which the curd is broken into smaller pieces before being placed in a cheese press.

Mold-ripened cheese

A cheese upon whose surface (and/or interior) a mold is encouraged to grow. Two types of mold are most common in cheese making. They are blue mold for blue cheeses and white mold for Camembert and related cheeses.

Molding

A step in cheese making during which the curd is placed in a cheese mold. The cheese mold will help produce the final shape of the cheese and aids in drainage.

Pasteurization

The heating of milk to destroy pathogenic organisms which may be harmful to man.

Pressing

A step in cheese making during which the curds are placed in a cheesecloth-lined mold and placed under pressure to remove more whey.

Raw milk

Milk which is taken fresh from the animal and has not been pasteurized.

Rennet

Rennets are enzymes of animal or vegetative origin. The rennet has the ability to coagulate milk. Animal rennet was originally extracted from the fourth stomach of the calf. Rennets are available in liquid and dried form.

Renneting

A step in cheese making in which rennet is added to milk in order to induce coagulation.

Ripening

A step in cheese making in which the milk is allowed to undergo an increase in acidity due to the activity of cheese starter culture bacteria.

Salting


A step in cheese making in which coarse flake salt is added to the curds before molding or to the surface of the finished cheese.

Whey

The liquid portion of milk which develops after coagulation of the milk protein. Whey contains water, milk sugar, albuminous proteins, and minerals.

White mold

A white mold (*penicillium candidum*) which is encouraged to grow on a number of soft cheeses in order to develop a pungent flavor. Camembert is perhaps the most famous of these cheeses.

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